

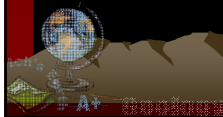
## What is a mineral? 4000 minerals on Earth !!!!

A naturally occurring, inorganic solid crystalline substance with a definite chemical composition.

A mineral must have these 5 major aspects:

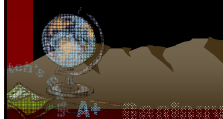
If even just one of these requirements are not satisfied, then the substance is **NOT** a mineral.

Naturally occurring  
Inorganic  
Solid  
Crystal  
Definite Chemical composition



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## What do these products have in common?



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**Language Arts:** Read "Ira Sleeps Over." Let students bring PJs & toothpaste.  
**Social Studies:** What was used before toothpaste was "invented"?



## Dig A Little Deeper A Bright Smile

**From Toothpaste and Minerals**

Toothpaste cleans your teeth and keeps them healthy.

**Science:** What minerals are found in toothpaste. Read about or research fluorite. Compare fluoride content in various brands.

**Math:** Survey class on brands used, chart or graph. **Health:** Discuss dental hygiene & special ingredients. **P.E.:** Stomp & squirt contest, use toothpaste & butcher paper.

## Naturally Occurring

**Naturally Occurring**

- Something formed in nature, not man-made.
- Example: Diamond, Copper, Salt

Minerals are a girl's best friend !



- Diamond is a mineral
- Cubic Zirconia is not

# WHY ?





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## Inorganic

### Inorganic

- Something that was never living.
- An inorganic substance was formed by earth processes.

Is ice inorganic?

Is paper inorganic?

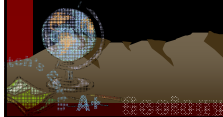
Is rock from magma inorganic?

Is a dead cat inorganic?



**Are synthetically created substance minerals?**

After all, they were never alive ????



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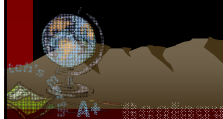
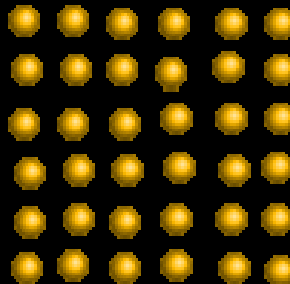
## Solid

### Solid

- Something that has a definite, ridged shape and volume.
- Gas and liquids are not minerals.

Is air a mineral?

Is mercury a mineral?



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## Crystalline Substance



## Definite chemical composition

### Definite chemical composition

- Every mineral has its own **chemical composition**.
- A chemical composition is like a **recipe for baking**.
- A mineral has **certain elements** combined in **certain amounts**.

Na + Cl = Salt  
Si + O<sub>2</sub> = Quartz

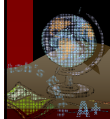
Just like baking,  
you must have the  
ingredients in just  
the right amounts.

Si + O<sub>2</sub> is not the  
same as Si + O<sub>3</sub>

1 egg  
1 cup of flour  
2 cups sugar

... will make  
something totally  
different than....

5 eggs  
1 cup flour  
8 cups sugar

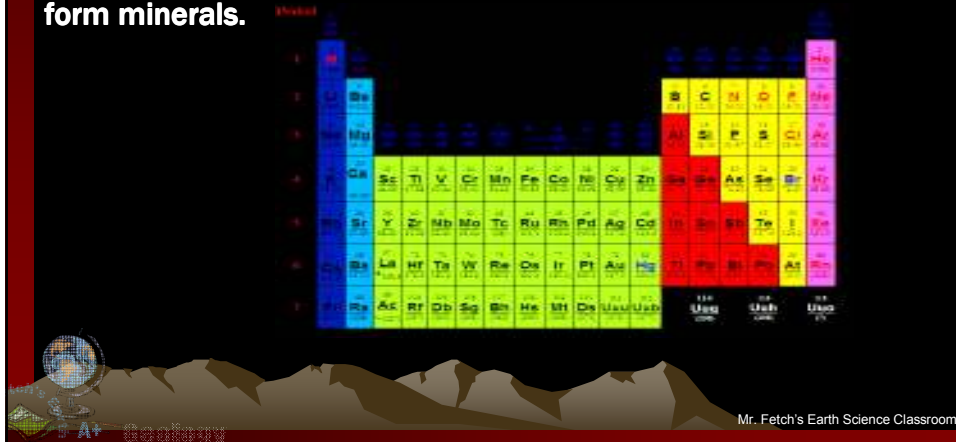


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## What are minerals made of?

Most minerals are made up of a combination of two or more elements.

Elements such as Na, Cl, Fe, C, O, Si – and others that you might recognize from the periodic table of elements combine together to form minerals.



## How do minerals form?

Minerals form through processes called crystallization.

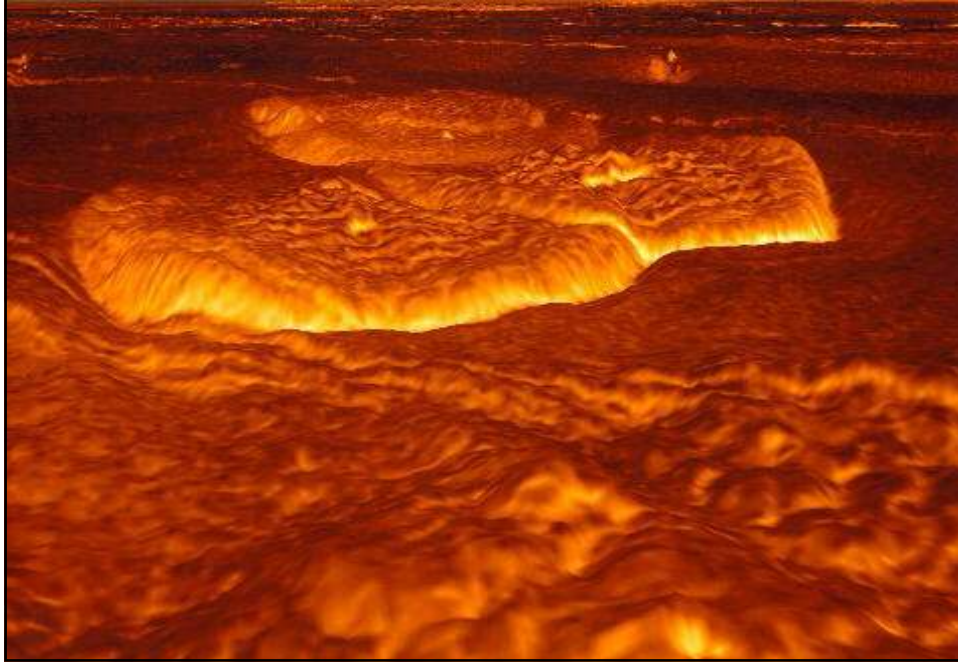
Crystallization is a “crystal-forming process.” As you know, minerals are crystals!

Minerals (crystals) can form in two ways:

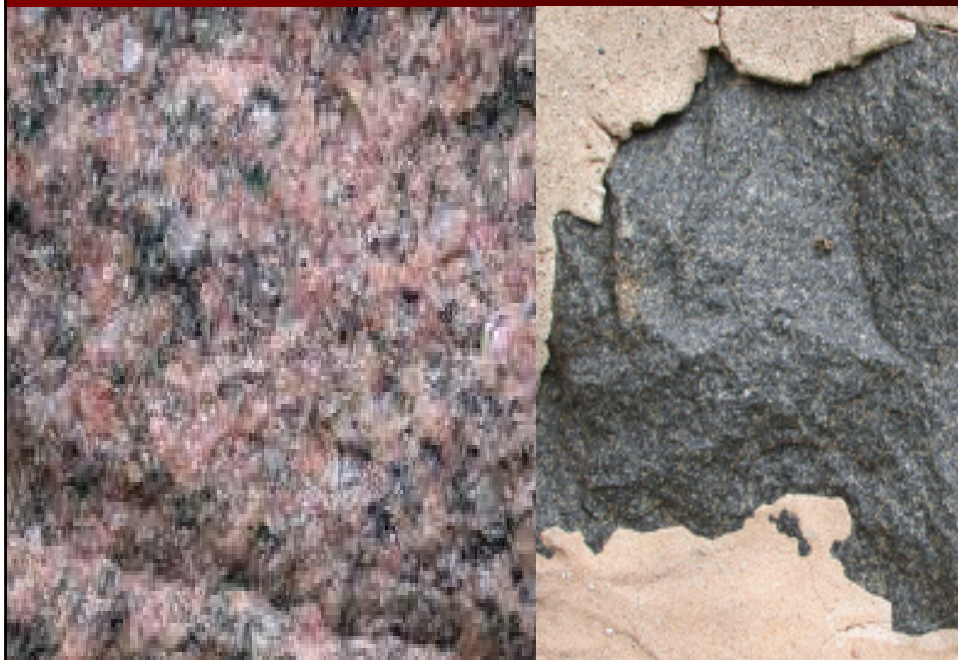
1. Cooling of Magma
2. Solution Evaporation



## Cooling of Magma



## Cooling of Magma



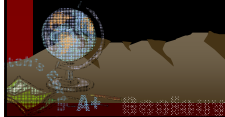
## Solution Evaporation

Minerals dissolve in water and create a solution.

Eventually, the water will evaporate and the minerals will fall out of solution and be left behind.

When minerals fall out of a solution they are said to precipitate.

Ex: Salt Water evaporating with salt crystals left behind.



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## Identifying Minerals...

**Hardness**

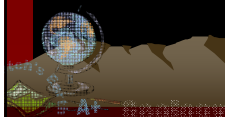
**Fracture**

**Color**

**Cleavage**

**Luster**

**Streak**



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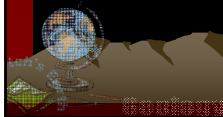


## Identification of Minerals

- What if you have two minerals that look exactly alike?
- How will you be able to tell one from the other?  
Study their characteristics.

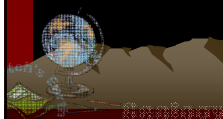
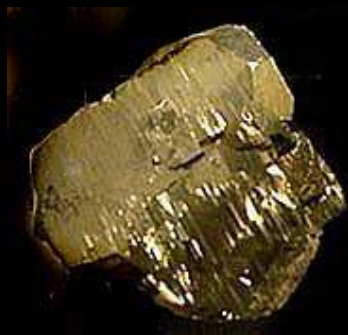
There are a number of different **properties** or characteristics that can give you clues to recognize different minerals.

**Those properties include: Color, luster, streak, hardness, cleavage, fracture, specific gravity, and a number of useful others.**



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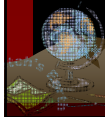
## Identification of Minerals: Which is Gold?



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## Identification of Minerals: There are over 4000 !



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## Color

Although color is an obvious feature of a mineral, it is often unreliable for identification.

Slight impurities can contaminate the mineral changing its color.

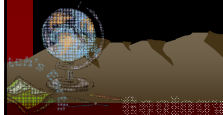


Example: Quartz can be white, pink, purple.



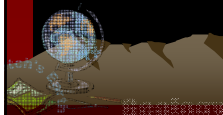
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## Color



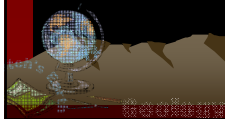
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## Color



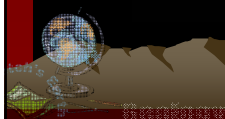
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## Color



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## Color



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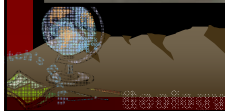
## Luster

Luster is the way a mineral reflects light.  
It is often described as either metallic or nonmetallic.

**METALLIC LUSTER**

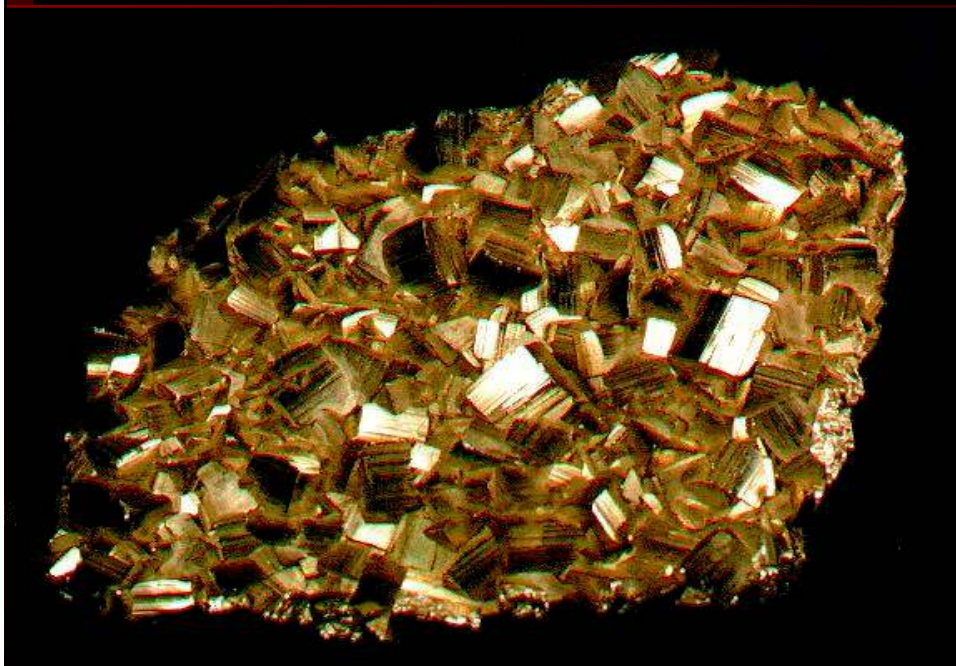


**NON-METALLIC LUSTER**



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## Luster: What luster is it?



**Luster: What luster is it?**



**Luster: What luster is it?**

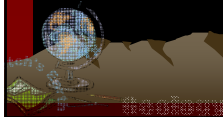


## Streak

If you were to scratch a mineral against a hard surface, like a porcelain tile it would leave behind a streak of colored powder.



Scientists use a streak plate to do the streak test.



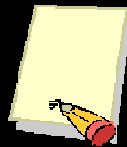
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## Streak

While a mineral's color may change, the color of its streak usually does not. Streak is often a much more helpful way to use color to identify a mineral.

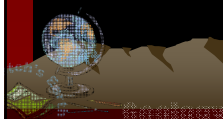
Where do you use streak in your everyday day life?

Writing with a pencil on paper.



Which mineral do you use to leave the streak?

Graphite



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## Streak



## Streak

Oolitic  
hematite



Specular  
hematite



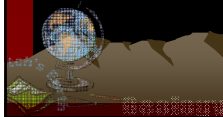


## Hardness

The measure of how easily a mineral can be scratched is known as its hardness.

Hardness of a mineral has nothing to do with whether it breaks easily or not.

Hardness is measured by using The Moh's Hardness Scale, which is a scale that ranks ten common minerals hardness.



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## Hardness

The scale is from 1-10.

Talc is the softest mineral and diamond is the hardest mineral.

**Moh's Scale of Hardness of Minerals**

		Common Items Hardness
1	Talc	
2	Gypsum	Fingernail = 2-2.5
3	Calcite	Copper coin = 3.5
4	Fluorite	Iron Nail = 4.5
5	Apatite	Glass = 5-5.5
6	Orthoclase	Streak plate = 6.5-7
7	Quartz	
8	Topaz	
9	Corundum	
10	Diamond	

2	Gypsum	Fingernail =	2-2.5
3	Calcite	Copper coin =	3.5
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5	Apatite	Glass =	5-5.5
6	Orthoclase	Streak plate =	6.5-7
7	Quartz		
8	Topaz		
9	Corundum		
10	Diamond		

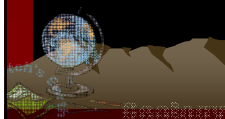
## Cleavage

The way a mineral **breaks apart** is another way that is helpful in identifying it.

Sometimes when you break a mineral, it will break along **flat, smooth** surfaces.



This results in a nice **clean cuts**



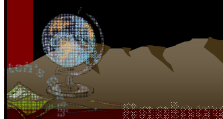
## Cleavage



## Cleavage

This is called **cleavage**.

Think of the way a sharp meat cleaver cuts meat. Nice, clean cuts.



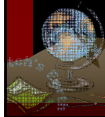
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## Fracture

Sometimes, minerals do not break in nice clean, flat cuts.

Minerals that break along flat, smooth surfaces are said to fracture.

The way a mineral breaks depends on the strength and arrangement of atomic bonds within the mineral.



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## Cleavage and Fracture



## Cleavage and Fracture

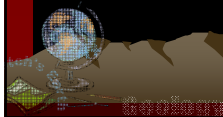
### MINERALS WITH CLEAVAGE

Mica  
Calcite  
Halite  
Fluorite

### MINERALS THAT FRACTURE

Quartz  
Olivine

All minerals fracture to some extent... even those with cleavage do not break PERFECTLY all the time.



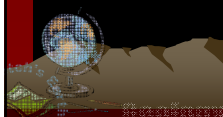
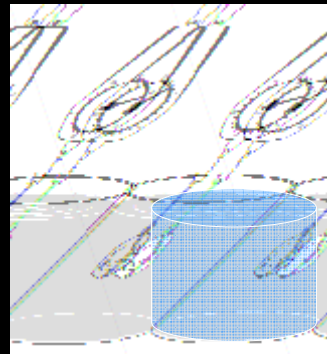
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## Specific Gravity

Minerals can be identified by comparing the weights of equal samples.

The specific gravity of a mineral is the ratio of its weight compared to the weight of an equal volume of water.

$$SG = \frac{\text{Weight of mineral}}{\text{Weight in equal volume of water}}$$



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## Specific Gravity

Gold has a specific gravity of **19**.

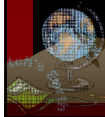
Pyrite has a specific gravity of **5**.

That means the gold is **19** times heavier than water and pyrite is **5** heavier than water.

If gold and pyrite look the same, you can tell them apart by finding their **specific gravity**.



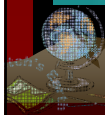
If you were to **heft** or lift two different minerals, the one with the **higher** specific gravity would feel **heavier**.



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## Magnetism

**Magnetism:** Some minerals are attracted to **magnets and metals**.



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## Acid Test

Acid Test: Weak HCl acid on carbonate minerals (those with  $\text{CO}_2$ ) will produce a chemical reaction.

$\text{CO}_2$  is given off as bubbles.



## Smell

Smell: Some minerals have a peculiar smell. Ex: sulfur

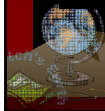
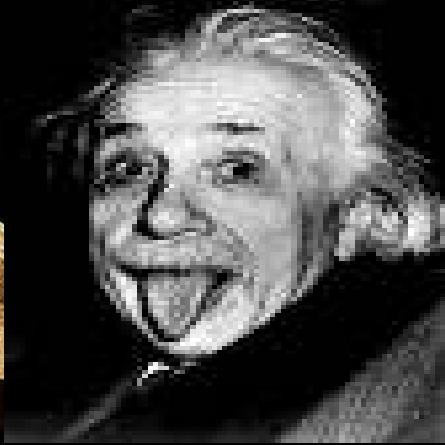




## Taste

Smell: Some minerals have a peculiar taste. Ex: halite

Halite



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## Touch

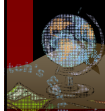
Smell: Some minerals have a peculiar feel. Ex: talc or graphite



SILKY



SOAPY

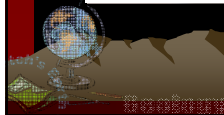
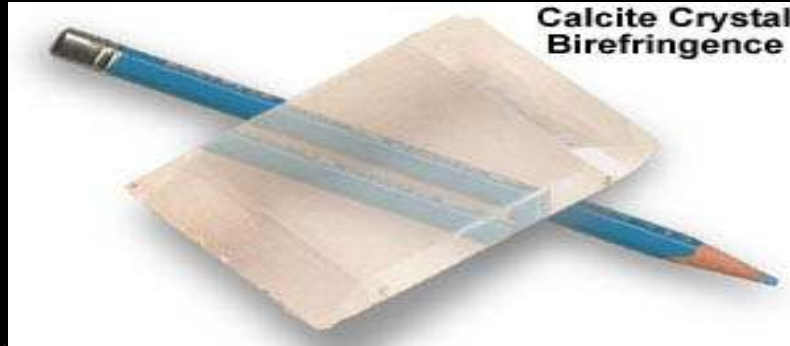


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## Refraction

**Double Refraction:** Light enters the mineral as one beam of light, but then it separates into to 2 beams.

You then see DOUBLE. Ex: calcite

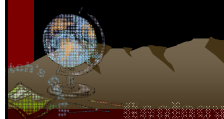
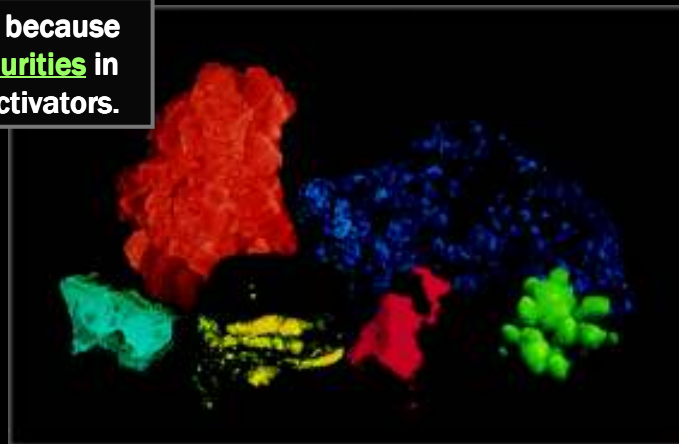


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## Fluorescence

**Fluorescence:** Some minerals glow brightly under a black light.

Minerals glow because they have impurities in them called activators.



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