Getting To Know the Periodic Table

Objective: To be able to use the periodic table to identify and classify elements and to use the periodic table to predict the behavior of elements

Procedure:

- 1. Number the groups.
- 2. Number the periods
- 3. Draw a heavy lack line between the metals and nonmetals.
- 4. Write the name of each of the following groups above the number:
 - Group 1 alkali metals
 - Group 2 alkaline earth metal
 - Group 3-12 (collectively) transition metals
 - Group 16 chalcogens Group 17 halogens Group 18 Noble gases
- 5. Write the names of the two rows at the bottom of the chart: lanthanides and actinides
- 6. Write the symbol of each element that exists as a gas at ordinary conditions in RED.
- 7. Write the symbol of each element that is a solid at ordinary conditions in BLACK.
- 8. Write the symbol of each element that is a liquid at ordinary condition in BLUE.
- 9. Write the symbol of each element that is a man-made element as an outline. Example: Pm
- 10. Place the atomic number for each element above the symbol.
- 11. Use the following chart to color the periodic table.

Halogen blue
Noble gases yellow
Alkali metals purple
Alkaline earth metals red
Transition elements green

Chalcogens brown
Lanthanides orange
Actinides light blue

12. Outline the symbol's box in dark green if it is RADIOACTIVE in its most common form.

Periodic Table Worksheet

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KEY:

RED {GAS}

H, He, N, O, F, Ne, Cl, Ar, Kr, Xe, Rn, Sm, Eu, Gd

BLUE {LIQUID}

Ga, Br, Cs, Hg, Fr

GREEN {RADIOACTIVE}

Tc, Po, At, Rn, Fr, Ra, Rf, Ha, La, Ce, Pr, Nd, Pm, Ac, Th, Pa, U, Np, Pu, Am, Cm, Bk, Cf, Es, Fm, Md, No, Lr

OUTLINE {SYNTHETIC}

Tc, Pm, Np, Pu, Am, Cm, Bk, Cr, Es, Fm, Md, No, Lr

BLACK {SOLID}

Li, Be, B, C, Na, Mg, Al, Si, P, S, K, Ca, Sc, Ti, V, Cr, Mn, Fe, Co, Ni, Cu, Zn, Ge, As, Se, Rb, Sr, Y, Zr, Nb, Mo, Ru, Rh, Pd, Ag, Cd, In, Sn, Sb, Te, I, Ba, Hf, Ta, W, Re, Os, Ir, Pt, Au, Tl, Pb, Bi, Po, At, Ra, Rf, Ha, Sg, Bh, Hs, Mt, Tb, Dy, Ho, Er, Tm, Yb, Lu, Uun, Uuu, Uub